Role of conditional entropy in experiments of Landauer principle

D. Chiuchiú¹, M. C. Diamantini^{1,2}, L. Gammaitoni^{1,2}

¹NiPS Laboratory - UNIPG ²INFN sezione di Perugia

Landauer Summer School - Perugia - 18/7/14





Perugia — 18/7/14

1 Introduction

2 Conditional Entropy

3 Gaussian Example

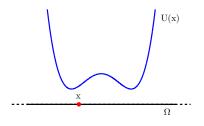
Davide Chiuchiú (UNIPG)

Perugia — 18/7/14

Landauer Summer School 2014 2 / 10

æ

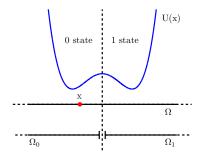
Bit Coding General Framework for 1-Dim Systems



- 1-Dimensional system
 - Each x value is a microstate
 - Ω is the set of all possible microstates
 - U(x) bistable and symmetric potential

Bit Coding

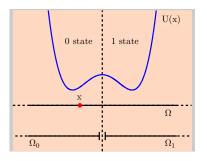
General Framework for 1-Dim Systems



- 1-Dimensional system
 - Each x value is a microstate
 - Ω is the set of all possible microstates
- U(x) bistable and symmetric potential
- Subdivide Ω in Ω_0 and Ω_1
- $x\in\Omega_0,\Omega_1\equiv$ bit in the 0,1 state

Bit Coding

General Framework for 1-Dim Systems



- 1-Dimensional system
- Each x value is a microstate
- Ω is the set of all possible microstates
- U(x) bistable and symmetric potential
- Subdivide Ω in Ω_0 and Ω_1
- $x\in\Omega_0,\Omega_1\equiv$ bit in the 0,1 state
- Heath bath a temperature T

x fluctuates \Rightarrow We describe its stochastic properties through a Probability Density Function P(x).

We than define the probabilities of being in the 0-1 logic states

$$P_0 = \int_{\Omega_0} P(x) dx, \quad P_1 = \int_{\Omega_1} P(x) dx$$

Entropies and Landauer Principle

Gibbs thermodynamical entropy

$$S_G = -K_B \int_{\Omega} P(x) \log P(x) dx$$

Shannon information entropy

$$S_S = -K_B \sum_{i=0,1} P_i \log P_i$$

Landauer Principle

Computation is a physical transformation that changes S_G and S_S . Heat production for this transformation obeys Clausius theorem

 $Q \geq -T\Delta S$

Davide Chiuchiú (UNIPG)

Entropies and Landauer Principle

Gibbs thermodynamical entropy

$$S_G = -K_B \int_{\Omega} P(x) \log P(x) dx$$

Shannon information entropy

$$S_S = -K_B \sum_{i=0,1} P_i \log P_i$$

Landauer Principle

Computation is a physical transformation that changes S_G and S_S . Heat production for this transformation obeys Clausius theorem

$$Q \ge -T\Delta S_G$$

$$\Delta S_G = \Delta S_S + \Delta S_{cond}$$
(1)

Warning1: Up to now $\Delta S_{cond} = 0$ was used with no clear justification. Warning2: There is no simple representation of S_{cond} and ΔS_{cond} .

(1) T. Sagawa J. Stat. Mech. (2014) P03025

For a system with bistable U(x), it is reasonable that, in experiments, non-equilibrium P(x) are functions with two peaks. We write

$$P(x) = P_a \eta_a(x) + P_b \eta_b(x)$$

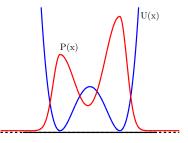
 η_a , η_b properties:

- peak functions.
- thay are non-negative with supports Ω_a , Ω_b respectively.

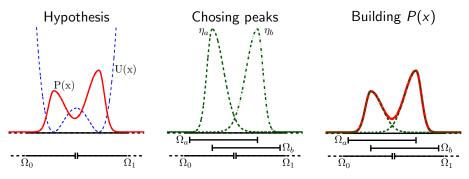
•
$$\int_{\Omega_a} \eta_a dx = \int_{\Omega_b} \eta_b dx = 1$$

Thanks to this last one

- $P_a + P_b = 1$
- *P_a* and *P_b* are the probability that a microstate *x* belongs to *η_a* or *η_b*



The sample structure of P(x) - [2]



Note that:

- Each peak has its own shape;
- Peaks significatively overlap near the boundary between Ω₀, Ω₁;
- Ω_a and Ω_b may not coincide with Ω_0 and $\Omega_1 \Rightarrow (P_a, P_b)$ is not (P_0, P_1) .

$$P_{0} = P_{a} \int_{\Omega_{a} \cap \Omega_{0}} \eta_{a} dx + P_{b} \int_{\Omega_{b} \cap \Omega_{0}} \eta_{b} dx$$
$$P_{1} = P_{a} \int_{\Omega_{a} \cap \Omega_{1}} \eta_{a} dx + P_{b} \int_{\Omega_{b} \cap \Omega_{1}} \eta_{b} dx$$

A simple formula for S_{cond}

$$S_{cond} = S_G - S_S = S_{ex} + S_{sh} + S_{ov}$$

 $S_{ex} = -K_B P_a \log P_a - K_B P_b \log P_b + K_B P_0 \log P_0 + K_B P_1 \log P_1$ $S_{ex} \text{ is the entropic measure of the error committed by exchanging } (P_a, P_b) \text{ with } (P_0, P_1).$

$$S_{sh} = P_a S_a + P_b S_b$$

 $S_a = -K_B \int_{\Omega_a} \eta_a \log \eta_a dx$
 S_{sh} gives the entropic measure of η_a and η_b shapes

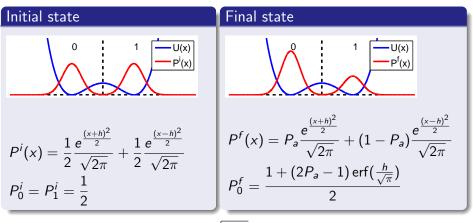


$$S_{ov} = P_a I(\eta_a, \eta_b, \frac{P_b}{P_a}) + P_b I(\eta_b, \eta_a, \frac{P_a}{P_b})$$
$$\frac{I(\eta_a, \eta_b, q)}{K_B} = -\int_{\Omega_a \cap \Omega_b} \eta_a \log\left(1 + q \frac{\eta_b}{\eta_a}\right) dx$$
$$S_{ov} \text{ is the entropic measure of } \eta_a$$
and η_b overlap in $P(x)$

η

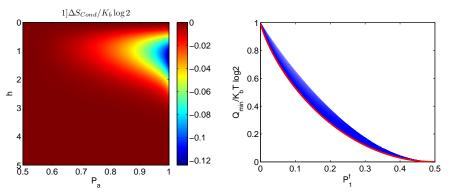
Gaussian Example

We reset a bit of information



Free parameters: P_a and $h \ (h \approx \sqrt{\frac{\Delta U}{K_B T}})$

Some results



- $\Delta S_{cond} \neq 0$
- for $h \approx 1 \left[\frac{\Delta U}{K_B T} \approx 1\right]$, ΔS_{cond} can become up to 25% of ΔS_G .
- ΔS_S is insufficient to characterize minimum heat production

- Brief introduction to ΔS_{cond} , writing it in a simple and intuitive way for bistable systems.
- Discussed the implications of ΔS_{cond} to minimum heat production with a simple example based on gaussian peaks.

Further readings

- D. Chiuchiú, M. C. Diamantini, L. Gammaitoni. *Role of conditional* entropy in experimental tests of Landauer Principle., arXiv:1406.2562
- T. Sagawa *Thermodynamic and Logical Reversibilities Revisited* J. Stat. Mech. (2014) P03025

< ロ > < 同 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < 回 > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ > < □ >